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INFORMATION DISCLOSURE STATEMENT BY APPLICANT <i>(Use as many sheets as necessary)</i>		Application Number	10/710,686
		Filing Date	July 28, 2004
		First Named Inventor	Phillip W. BARAK
		Art Unit	1723
		Examiner Name	
Sheet	1	of	3
		Attorney Docket Number	054030-0062

U. S. PATENT DOCUMENTS

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NON PATENT LITERATURE DOCUMENTS			
Examiner Initials*	Cite No. ¹	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc.), date, page(s), volume-issue number(s), publisher, city and/or country where published.	T ²
	1.	Abbona, F., et al., "Synthetic struvite, MgNH ₄ PO ₄ ·6H ₂ O: Correct polarity and surface features of some complementary forms," <i>Acta Cryst.</i> B40:223-227 (1984).	
	2.	Abbona, F., et al., <i>J. Cryst. Growth</i> 54:6-14 (1982).	
	3.	Barak, P. 1990. SPECIES: A spreadsheet program for modeling speciation of soil solution. <i>J. Agron. Educ.</i> 19:44-46. (http://www.soils.wisc.edu/~barak/species.htm)	
	4.	CEEP, Comite European d'Etudes des Polyphosphates. Conference Summary; Second Int'l Conf. on the recovery of phosphorus from sewage and animal wastes. Noordwijkerhout, The Ne	
	5.	Gonzalez-Munoz, M.T., et al., "Struvite and calcite crystallization induced by cellular membranes of <i>Myxococcus xanthus</i> ," <i>J. Cryst. Growth</i> 163:434-439 (1996).	
	6.	Jaffer, Y., et al., "Potential phosphorus recovery by struvite formation," <i>Water Research</i> 36:1834-1842 (2002).	
	7.	Liberti, L., D., et al., "REM NUT ion exchange plus struvite precipitation process." <i>Environmental Technology</i> 22:1313-1324 (2001).	
	8.	Mann, S., et al., "Crystallization at inorganic-organic interfaces-biominerals and biomimetic synthesis" <i>Science</i> 261:1286-1292 (1993).	
	9.	Mann, S., et al., "Controlled crystallization of CaCO ₃ under stearic-acid monolayers." <i>Nature</i> 334:692-695 (1988).	
	10.	Ohlinger, K.N., et al., "Kinetics effects on preferential struvite accumulation in wastewater," <i>J. Environ. Eng.</i> 125:730-737 (1999).	

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	11.	Pierzynski, G.M., J.T. Sims., and G.F. Vance. 2000. Soils and Environmental Quality. 2nd ed. CRC Press, Boca Raton, London, New York, Washington D.C.			T ²
	12.	Rajam, S., et al., "Oriented crystallization of CaCO ₃ under compressed monolayers. Part 1.--Morphological studies of mature crystals," J. Chem. Soc.-Faraday Trans. 87:727-734			
	13.	Rawn, A.M., et al., "Multiple stage sewage digestion," Trans. Am. Soc. Civil Eng. 105:93-132 (1939).			
	14.	Smith, R.M., and Martell, A.E., "NIST critically selected stability constants of metal complexes," version 6.0. U.S. Dept. of Commerce, National Institute of Standards and Tec			
	15.	Tantinirundr, U. "Phosphorus removal in an aerobic supernatant by struvite crystallization without addition of chemicals." Master's thesis, University of Wisconsin, Madison (2			
	16.	Taylor, A.W., et al., "Solubility products of magnesium ammonium and magnesium potassium phosphates," Trans. Faraday Soc. 5(9):1585-1589 (1963).			

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